

*trans*-Tetrachlorobis(triphenylphosphine)osmium(IV)Zhengkai Li<sup>a\*</sup> and Zhongyuan Zhou<sup>b</sup><sup>a</sup>The Faculty of Chemistry, The Sichuan University, Chengdu, Sichuan, People's Republic of China, and <sup>b</sup>Department of Applied Biology and Chemical Technology, The Hong Kong Polytechnic University, Hong Kong, China

Correspondence e-mail: bczyzhou@inet.polyu.edu.hk

## Key indicators

Single-crystal X-ray study

T = 294 K

Mean  $\sigma(\text{C}-\text{C}) = 0.008 \text{ \AA}$ 

R factor = 0.032

wR factor = 0.078

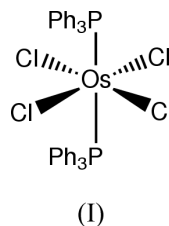
Data-to-parameter ratio = 19.8

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The structure of the title compound,  $[\text{OsCl}_4(\text{C}_{18}\text{H}_{15}\text{P})_2]$ , is centrosymmetric. The geometry of the  $\text{Os}^{\text{IV}}$  center is octahedral with the P atoms occupying mutually *trans*-positions.

## Comment

Osmium(IV) complexes of the type *trans*- $[\text{Os}(\text{PR}_3)_2\text{X}_4]$  ( $\text{X} = \text{Cl}$  or  $\text{Br}$ ) have been characterized spectroscopically (Salmon & Walton, 1978; Cipriano *et al.*, 1990), but only the  $\text{Os}^{\text{IV}}(\text{P}-\text{Me}_2\text{Ph})_2\text{Cl}_4$  species has been structurally characterized (Aslanov *et al.*, 1970). In our preparation of  $\text{OsCl}_2(\text{PPh}_3)_3$  from the reaction of  $\text{K}_2\text{OsCl}_6$  with  $\text{PPh}_3$  according to the literature method of Hoffmann & Caulton (1975), the title compound, (I), was also formed and subsequently characterized crystallographically.



The structure of (I) (Fig. 1) is centrosymmetric. The geometry around the  $\text{Os}^{\text{IV}}$  center is octahedral with two P atoms in mutually *trans* positions. The  $\text{Os}\cdots\text{Cl}$  distances [mean  $2.313(6) \text{ \AA}$ ] are experimentally equivalent to those reported for the  $\text{Os}^{\text{IV}}(\text{PMe}_2\text{Ph})_2\text{Cl}_4$  structure [mean  $2.319(3) \text{ \AA}$ ; Aslanov *et al.*, 1970], but significantly shorter than those for  $[\text{Os}^{\text{III}}(\text{PEt}_3)_2\text{Cl}_4]^-$  [mean  $2.380(2) \text{ \AA}$ ; Kim *et al.*, 1984], as expected.

## Experimental

The title compound was obtained as a by-product when  $\text{Os}(\text{PPh}_3)_3\text{Cl}_2$  was prepared according to the literature method (Hoffmann & Caulton, 1975). A mixture of  $\text{K}_2\text{OsCl}_6$  and  $\text{PPh}_3$  (1:7) in 2-propanol and  $\text{H}_2\text{O}$  (2:1) was refluxed for 4 h to give a green precipitate, which was collected and extracted with benzene. A small amount of benzene insoluble orange-red solid was collected by filtration and recrystallized from a  $\text{CH}_2\text{Cl}_2$  solution layered with hexane to afford orange-red crystals suitable for X-ray structure analysis.

Crystal data

[OsCl<sub>4</sub>(C<sub>18</sub>H<sub>15</sub>P)<sub>2</sub>]  
*M<sub>r</sub>* = 856.54  
 Monoclinic, *P*2<sub>1</sub>/*n*  
*a* = 9.3364 (9) Å  
*b* = 20.948 (2) Å  
*c* = 9.6251 (9) Å  
 β = 117.425 (2)°  
*V* = 1670.9 (3) Å<sup>3</sup>  
*Z* = 2

*D<sub>x</sub>* = 1.702 Mg m<sup>-3</sup>  
 Mo *K*α radiation  
 Cell parameters from 3136 reflections  
 θ = 1.0–27.5°  
 μ = 4.26 mm<sup>-1</sup>  
*T* = 294 (2) K  
 Prism, orange-red  
 0.20 × 0.16 × 0.10 mm

Data collection

Bruker CCD area-detector diffractometer  
 φ and ω scans  
 Absorption correction: empirical (SADABS; Sheldrick, 1996)  
*T<sub>min</sub>* = 0.483, *T<sub>max</sub>* = 0.676  
 11 100 measured reflections

3844 independent reflections  
 2867 reflections with *I* > 2σ(*I*)  
*R<sub>int</sub>* = 0.044  
 θ<sub>max</sub> = 27.6°  
*h* = -6 → 12  
*k* = -25 → 27  
*l* = -12 → 12

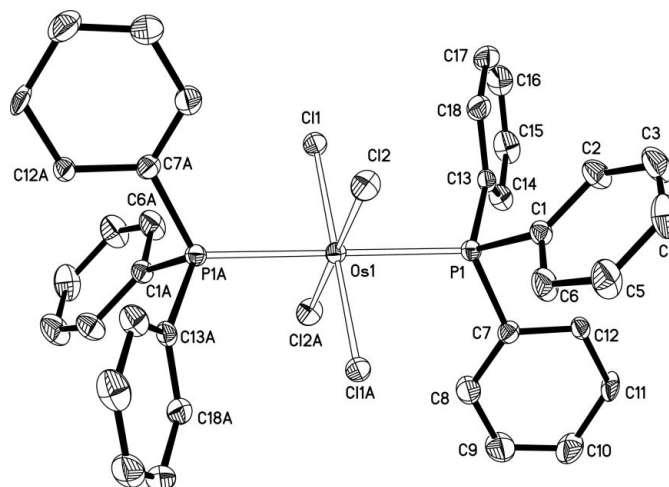
Refinement

Refinement on *F*<sup>2</sup>  
*R* [*F*<sup>2</sup> > 2σ(*F*<sup>2</sup>)] = 0.032  
*wR* (*F*<sup>2</sup>) = 0.078  
*S* = 0.98  
 3844 reflections  
 194 parameters

H-atom parameters constrained  
*w* = 1/[σ<sup>2</sup>(*F<sub>o</sub>*<sup>2</sup>) + (0.0409*P*)<sup>2</sup>]  
 where *P* = (*F<sub>o</sub>*<sup>2</sup> + 2*F<sub>c</sub>*<sup>2</sup>)/3  
 (Δ/σ)<sub>max</sub> < 0.001  
 Δρ<sub>max</sub> = 0.98 e Å<sup>-3</sup>  
 Δρ<sub>min</sub> = -0.60 e Å<sup>-3</sup>

The C-bound H atoms were placed in their geometrically calculated positions and included in the final refinement in the riding model approximation.

Data collection: Bruker *SMART* (Bruker, 1999); cell refinement: Bruker *SMART*; data reduction: Bruker *SHELXTL-NT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: Bruker *SHELXTL-NT*; software used to prepare material for publication: Bruker *SHELXTL-NT*.



**Figure 1**  
 Molecular structure of (I) with displacement ellipsoids shown at the 30% probability level (*SHELXTL-NT*; Bruker, 1999). Symmetry operation for atoms labeled with A: -*x*, 1 - *y*, -*z*.

References

Aslanov, L., Mason, R., Wheeler, A. G. & Whimp, P. O. (1970). *Chem. Commun.* pp. 30–31.  
 Bruker (1999). *SHELXTL-NT*. Version 5.10. Bruker AXS Inc., Madison, Wisconsin, USA.  
 Cipriano, R. A., Levason, W., Mould, R. A. S., Pletcher, D. & Webster, M. (1990). *J. Chem. Soc. Dalton Trans.* pp. 339–347.  
 Hoffmann, P. R. & Caulton, K. G. (1975). *J. Am. Chem. Soc.* **97**, 4221–4228.  
 Kim, E. E., Eriks, K. & Magnuson, R. (1984). *Inorg. Chem.* **23**, 393–397.  
 Salmon, D. J. & Walton, R. A. (1978). *Inorg. Chem.* **17**, 2379–2382.  
 Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.  
 Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.